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# **Determination of Thermodynamic Properties via Partition Function Approaches Derived** from the Non-Relativistic Schrödinger Equation

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#### ABSTRACT

A critical challenge in molecular thermodynamics is accurately predicting how variations in dimensionless potential parameters influence macroscopic properties such as magnetisation, specific heat capacity, and molecular vibrational entropy. Traditional approaches often fail to fully capture these dependencies, limiting their predictive accuracy. In this study, we address this gap by developing a rigorous framework based on partition function formulations derived directly from the non-relativistic Schrödinger equation. Our main aim is to investigate the impact of dimensionless potential parameters ( $V_0$  and  $V_1$ ) and the temperature parameter (B) on energy spectrum and key thermodynamic and magnetic properties. We systematically compute the energy spectrum for a wide range of potential parameter values and analyse their effects on magnetisation, magnetic susceptibility, internal energy, free energy, entropy, and specific heat capacity. The findings reveal that energy spectrum exhibit strong dependence on these parameters, with  $V_0$  generally reducing energy levels while  $V_1$  enhances them. Magnetisation typically decreases with  $V_0$  but increases with  $V_1$  and higher temperatures, while magnetic susceptibility shows complex patterns of **Dimensionless Potential** enhancement and suppression. Vibrational thermodynamic properties, including internal energy and free energy, also display significant variations tied to the interplay of potential parameters. This study provides a robust, first-principles-Magnetic Susceptibility, based method for understanding and predicting how microscopic potential parameters govern macroscopic thermodynamic behaviour, advancing the fundamental knowledge and practical capabilities of molecular thermodynamics.

Partition Function, Specific Heat Capacity.

Keywords:

Parameter,

Magnetisation,

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## **INTRODUCTION**

Understanding forecasting and thermodynamic properties is essential across disciplines like chemical engineering, materials science, energy physics and molecular physics. Key properties-including internal energy, entropy, free energy, and heat capacityprovide valuable information about the stability, chemical reactivity, and equilibrium characteristics of molecular systems (Abu-shady et al., 2023). Statistical mechanics offers one of the most robust and informative frameworks for evaluating these macroscopic properties, effectively linking quantum-level behaviour

with observable large-scale phenomena (Abu-shady et al., 2023).

Central to this approach is the partition function, a key element in statistical mechanics. It encapsulates how energy states are statistically distributed in a system and forms the basis for deriving all thermodynamic properties (Abu-shady et al., 2023). To compute the partition function, one must understand the energy levels of the system-information typically acquired for molecular systems by solving the Schrödinger equation, a core principle of quantum mechanics that governs the evolution of quantum states over time (Abu-shady *et al.*, 2023).

Specifically, the non-relativistic Schrödinger equation is often employed to model molecular systems in cases where relativistic effects are insignificant. Solving this equation yields the permissible energy levels of the system, which are essential for building the partition function. With the partition function determined, numerous thermodynamic properties can be derived through standard mathematical formulations (Manga *et al.*, 2023).

This method—referred to as partition function modelling grounded in quantum mechanics—enables the prediction of thermodynamic behaviour from first principles (Onate *et al.*, 2022). It is particularly advantageous for analysing isolated molecules, gases, and other systems where quantum effects are prominent (Mulian, 2025). Additionally, it facilitates predictions independent of experimental data, making it highly beneficial for both theoretical research and practical applications such as reaction kinetics, phase equilibrium analysis, and the design of new materials (Ding *et al.*, 2011).

Within this framework, the current research emphasizes the evaluation of thermodynamic properties through partition function models based on the non-relativistic Schrödinger equation. By examining both the theoretical foundations and computational methods, this study seeks to highlight the efficiency and relevance of this quantum-statistical approach in contemporary thermodynamic investigations.

2. Hylleraas plus Yukawa Potential (Modified Hylleraas Potential (MHsP))

The study of both relativistic and non-relativistic solutions to wave equations within quantum mechanical systems has gained increasing global interest (Umirzakov, 2019). Over the past decade, numerous researchers have developed mathematical frameworks to solve the Schrödinger equation, focusing on deriving both the energy spectrum and corresponding wavefunctions (Dong, 2000). These solutions have been applied across a wide range of physical systems. In this work, we utilize the Modified Hylleraas Potential (MHp), which allows for exact solutions with relatively low mathematical complexity (Reiss, 1980). Originally introduced to model intermolecular interactions, the Hylleraas potential has been shown to be well-suited for multi-particle systems and vibrational energy analyses of diatomic molecules. According to Ikot et al., (2021), the Hylleraas potential exhibits a multi-exponential parameter structure, yet limited research has explored its bound state solutions within both relativistic and nonrelativistic quantum systems. To further extend its applicability, a variant known as the Modified Hylleraas

plus Yukawa Potential (MHsP) has also been proposed for similar investigations (Manga *et al.*, 2021).

$$V(r) = \frac{V_0}{b} \frac{a - e^{-2\beta r}}{1 - e^{-2\beta r}} + \frac{V_1 e^{-2\beta r}}{r}$$
(1)

Here,  $\beta$  represents the potential screening parameter. Since the introduction of this potential by Hylleraas, various modifications have been suggested; however, this study concentrates specifically on the form given in equation (1). It is worth noting that only a limited number of researchers have explored or applied this potential since its initial proposal.

#### Radial Solution of the Schrödinger equation based on Hylleraas plus Yukawa potential (MHsP)

The radial motion of a particle under a spherically symmetric potential V(r) is described by the following expression (Ikot *et al.*, 2021: Lublinsky *et al.*, 2017)

$$\frac{d^{2}R}{dr^{2}} + \left[\frac{2\mu}{\hbar^{2}}\left(E - V(r)\right) - \frac{\delta}{r^{2}}\right]R(r) = 0 \qquad (2)$$
Where  $\delta = \tau(\tau + 1)$  equation (2) becomes
$$\frac{d^{2}R}{dr^{2}} + \left[\frac{2\mu}{\hbar^{2}}\left(E - V(r)\right) - \frac{\tau(\tau+1)}{r^{2}}\right]R(r) = 0 \qquad (3)$$

Analytical solutions to Equation (3) are not feasible when  $\tau \neq 0$  due to the complexity introduced by the centrifugal term (Manga *et al.*, 2021). To address this, we employ the Greene and Aldrich approximation method for the centrifugal term as outlined Edet *et al.*, 2023).

$$\frac{1}{r^2} = \frac{4\delta^2 e^{-2\delta r}}{\left(1 - e^{-2\delta r}\right)^2} = \frac{1}{r} = \frac{2\delta e^{-2\delta r}}{\left(1 - e^{-2\delta r}\right)}$$
(4)

It is important to note that for short-range potentials, Equation (4) serves as a reasonable approximation for  $\frac{1}{r^2}$ . However, this approximation becomes less accurate when the screening parameter  $\delta$  is large, making it unsuitable for describing the centrifugal barrier under such conditions. The approximation remains valid only when  $\delta r \ll 1$ . Therefore, by substituting Equation (1) into Equation (2), and subsequently applying the Greene and Aldrich approximation as defined in Equation (4), we obtain the following expression (Edet *et al.*, 2023).

$$\frac{\frac{d^2R}{dr^2} + \left[\frac{2\mu}{\hbar^2} \left(E - \frac{V_0}{b} \frac{a - e^{-2\gamma r}}{1 - e^{-2\gamma r}} + \frac{2\gamma V_1 e^{-4\gamma r}}{(1 - e^{-2\gamma r})}\right) - \frac{4\gamma^2 e^{-2\gamma r} \tau(\tau + 1)}{(1 - e^{-2\gamma r})^2} R(r) = 0$$
(5)

The coordinate in Equation (1) is transformed as follows.

$$\frac{d^{2}}{dr^{2}} + \frac{1}{v}\frac{d}{dr} + \frac{1}{v^{2}(1-v)^{2}} \left(\frac{\mu E}{2\gamma^{2}\hbar^{2}} - \frac{\mu Ev}{\gamma^{2}\hbar^{2}} + \frac{\mu Ev^{2}}{2\gamma^{2}\hbar^{2}} - \frac{V_{0}a}{4\gamma^{2}b} + \frac{V_{0}av}{4\gamma^{2}b} + \frac{V_{0}v^{2}}{4\gamma^{2}b} - \frac{V_{1}v}{\gamma} + \frac{V_{1}v^{2}}{\gamma} - v\tau^{2} - v\tau\right) R(v) = 0$$
(6)

A transformation is applied to Equation (6) to facilitate obtaining its hypergeometric solution.

$$\frac{d^2}{dr^2} + \frac{1(1-v)}{v(1-v)}\frac{d}{dr} + \frac{1}{v^2(1-v)^2}\{(-C_1 - C_3 + C_4)u^2 + (-2C_1 + C_3 + C_2 - C_4 - \tau^2 - \tau)v - C_1 - C_2\}$$
(7)

The following dimensionless constants, as defined in Equation (7), were used (Edet *et al.*, 2023).  $\frac{\mu E}{r} = E_{r,r} = C_{r,r} C_{r,r} = \frac{V_{0}a}{r} C_{r,r} = \frac{V_{0}}{r} C_{r,r} = \frac{V_{1}}{r}$ 

$$\frac{\mu E}{2\gamma^2 \hbar^2} = \mathcal{E}_{n,l} = C_1, C_2 = \frac{v_0 a}{4\gamma^2 b}, C_3 = \frac{v_0}{4\gamma^2 b}, C_4 = \frac{v_1}{\gamma},$$
(8)  
Equation (7) meets the criteria of a hypergeometric function in the following form (Ettah. 2021).

$$\psi_n'' + \frac{\chi_1 - \chi_2 s}{s(1 - \chi_3 s)} \psi' + \frac{1}{s^2 (1 - \chi_3 s)^2} [-\mathfrak{T}_1 s^2 + \mathfrak{T}_2 s - \mathfrak{T}_3] \psi(s) = 0$$
(9)
Comparing equation (7) and (8) we have

 $\mathfrak{T}_{1} = -C_{1} - C_{3} + C_{4}; \ \mathfrak{T}_{2} = -2C_{1} + C_{3} + C_{2} - C_{4} - \tau^{2} - \tau; \ \mathfrak{T}_{3} = -C_{1} - C_{2}$ (10) The Gauss hypergeometric equation can be expressed in the form (Edet *et al.*, 2023).

$$\delta = \frac{1}{2} \Big( (1 - \gamma_1) \pm \sqrt{(1 - \gamma_1)^2 + 4\mathfrak{T}_3} \Big)$$
(11)  

$$\delta^2 = -C_1 - C_2$$
(12)

The coiling velocity, determined using the Nikiforov-Uvarov functional analysis method, is expressed as follows (Jia *et al.*, 2012).

$$\omega = \frac{1}{2\gamma_3} \left( (\gamma_3 + \gamma_1 \gamma_3 - \gamma_2) \pm \sqrt{(\gamma_3 + \gamma_1 \gamma_3 - \gamma_2)^2 + 4\left(\frac{\mathfrak{X}_1}{\gamma_3} + \gamma_2 \mathfrak{X}_3 - \mathfrak{X}_2\right)} \right)$$
(13)  
$$\omega = \frac{1}{2} \left( 1 + \sqrt{1 + 4(-2C_3 + 2C_4 + \tau^2 + \tau - 2C_2)} \right)$$
(14)

After equating equation (12) by equation (14) we deduced our bound state energy eigenvalue equation as;  $P_{m} = C_{m} + \left(\frac{-C_{3}+C_{4}+C_{2}-(\omega+n)^{2}}{2}\right)^{2}$ (14)

$$E_{n,l} = C_2 + \left(\frac{-c_3 + c_4 + c_2 - (\omega + n)^2}{2(\omega + n)}\right)$$
(14)

Equation (15) can be rewrite by substituting the dimensionless constant as given in equation (8), we have

$$E_{n,\tau} = \frac{2\gamma^2 \hbar^2}{\mu} \frac{V_0 a}{4\gamma^2 b} + \frac{2\gamma^2 \hbar^2}{\mu} \left( \frac{-\frac{V_0}{4\gamma^2 b} + \frac{V_1 + \frac{V_0 a}{\gamma} - (\omega + n)^2}{\gamma}}{2(\omega + n)} \right)^2$$
(15)



Figure 1: Computed energy spectrum under varying potential parameters

Figure 1 shows the energy spectrum computed from the non-relativistic Schrödinger equation for different dimensionless potential parameters. In subplots (a) and (b), the energy levels are higher for lower values of  $V_0$  and  $V_1$  with sensitivity to variations in  $\lambda$  and  $V_1$ . As these parameters increase, the energy spectrum declines. A similar behaviour is observed in subplots (c) and (d), highlighting the strong dependence of energy eigenvalues on the potential parameters.

$$\begin{aligned} Partition function can be express as follows\\ PF &= \frac{1}{2\sqrt{-k1\beta}} e^{\frac{1}{2}(-2k1k2+k2k3)\beta - \frac{1}{2}\sqrt{-k1\beta}\sqrt{k2k3^2\beta}} \sqrt{\pi}(1 + e^{\sqrt{-k1\beta}\sqrt{k2k3^2\beta}}(-1 + \operatorname{Erf}[\frac{p^2\sqrt{-k1\beta}+\sqrt{k2k3^2\beta}}{2p}]) + \operatorname{Erf}[\frac{1}{2}p\sqrt{-k1\beta} - \frac{\sqrt{k2k3^2\beta}}{2p}]) \end{aligned}$$



Figure 2: The plot of Partition Function Under Varying Dimensionless Potential Parameters

(17)

For figure 2; At a fixed value of  $\lambda$ =0.005 for the dimensionless potential parameter (Dppr), results indicate that an increase in  $V_0$  leads to a decrease in the computed energy spectrum, whereas increasing  $V_1$  results in higher energy levels. This suggests that raising  $V_1$  elevates the energy spectrum. Similarly, when  $V_1$  is held constant at 0.005, the energy spectrum decreases with increasing  $V_0$  consistent with the trend shown in plot (a). Furthermore, at a constant  $V_0$  a higher

value of  $\lambda$  results in elevated energy levels, indicating that this dimensionless parameter also contributes to an increase in the energy spectrum. When  $\lambda$ =0.01, the energy spectrum continues to rise with increasing  $V_1$ , while larger values of  $V_0$  cause a slight reduction in the spectrum, resulting in a downward shift of the energy curves. At a fixed  $V_1 = 0.01$ , the spectrum increases with  $\lambda$ , while higher  $V_0$  values consistently correspond to lower energy levels across all  $\lambda$  values.

### Magnetic magnetisation can be express as follows

$$MM = \frac{P_1(P_2(\frac{P_3P_4}{\sqrt{\pi}} + \frac{P_5P_6}{\sqrt{\pi}} - P_7) + P_8 + P_9(P_{10}))}{P_2}$$

$$P_{1} = e^{-\frac{1}{2}(-\frac{f}{8}+\frac{u}{2})\beta + \frac{1}{8}\sqrt{-f\beta}\sqrt{u^{2}\beta}}\sqrt{-f\beta}$$
(18)

$$P_{2} = \frac{1}{\sqrt{-f\beta}} e^{\frac{1}{2}(-\frac{1}{3}+\frac{1}{2})\beta - \frac{1}{3}\sqrt{-f\beta}} \sqrt{2\pi}$$
(19)

$$P_{3} = 2e^{-\left(\frac{(2+\sqrt{1+4}(1-f+2u))\beta}{8\sqrt{2}} - \frac{\sqrt{-f\beta}}{\sqrt{2}(2+\sqrt{1+4}(1-f+2u))}\right)^{2}}$$
(20)  
$$P_{3} = \left(\frac{(2+\sqrt{1+4}(1-f+2u))\beta}{8\sqrt{2}} - \frac{\sqrt{-f\beta}}{\sqrt{2}\sqrt{2}\sqrt{2}\sqrt{u^{2}\beta}} - \frac{\sqrt{2}\sqrt{u^{2}\beta}}{\sqrt{2}\sqrt{u^{2}\beta}}\right)$$
(21)

$$P_{4} = \left(-\frac{(2+\sqrt{1+4}(1-f+2u))\beta}{16\sqrt{2}\sqrt{-f\beta}} - \frac{\sqrt{-f\beta}}{4\sqrt{2}\sqrt{1+4}(1-f+2u)} - \frac{\sqrt{2}\sqrt{u^{2}\beta}}{\sqrt{1+4}(1-f+2u)(2+\sqrt{1+4}(1-f+2u))^{2}}\right)$$
(21)

$$P_{5} = 2e^{\frac{1}{4}\sqrt{-f\beta}\sqrt{u^{2}\beta} - \frac{\left(\frac{(2+\sqrt{1+4}(1-f+2u))}{\sqrt{-f\beta}} + \frac{\sqrt{1-2}}{\sqrt{2}}\right)^{2}}{(2+\sqrt{1+4}(1-f+2u))^{2}}}$$
(22)

$$P_{6} = \left(\frac{\frac{(2+\sqrt{1+4}(1-f+2u))^{2}\beta}{16\sqrt{2}\sqrt{-f\beta}} \frac{(2+\sqrt{1+4}(1-f+2u))\sqrt{-f\beta}}{2\sqrt{2}\sqrt{1+4}(1-f+2u)}}{2+\sqrt{1+4}(1-f+2u)} + \frac{2\left(\frac{(2+\sqrt{1+4}(1-f+2u))^{2}\sqrt{-f\beta}}{8\sqrt{2}} + \frac{\sqrt{u^{2}\beta}}{\sqrt{2}}\right)}{\sqrt{1+4}(1-f+2u)\left(2+\sqrt{1+4}(1-f+2u)\right)^{2}}\right)$$
(23)

$$P_{7} = \frac{e^{\frac{1}{4}\sqrt{-f\beta}\sqrt{u^{2}\beta}} \sqrt{u^{2}\beta(-1+\text{Erf}[\frac{(2+\sqrt{1+4}(1-f+2u))^{2}\sqrt{-f\beta}}{8\sqrt{2}}+\sqrt{u^{2}\beta}]}}{8\sqrt{-f\beta}}}{8\sqrt{-f\beta}}$$
(24)

$$P_{8} = \frac{1}{(-f\beta)^{3/2}} e^{\frac{1}{2}(-\frac{f}{8}+\frac{u}{2})\beta - \frac{1}{8}\sqrt{-f\beta}\sqrt{u^{2}\beta}} \sqrt{\frac{\pi}{2}}\beta(1 + e^{\frac{1}{4}\sqrt{-f\beta}\sqrt{u^{2}\beta}}(-1 + e^{\frac{1}{4}\sqrt{-f\beta}\sqrt{u^{2}\beta}})) = e^{\frac{(2+\sqrt{1+4}(1-f+2u))^{2}\sqrt{-f\beta}}{2}} e^{\frac{(2+\sqrt{1+4}(1-f+2u))^{2}\sqrt{-f\beta}}{2}} e^{\frac{1}{2}(-\frac{1}{4}+\frac{u^{2}}{2})} e^{$$

$$\operatorname{Erf}\left[\frac{\frac{8\sqrt{2}}{2+\sqrt{1+4(1-f+2u)}}}{2+\sqrt{1+4(1-f+2u)}}\right] \operatorname{Erf}\left[\frac{(2+\sqrt{1+4(1-f+2u)})\sqrt{-f\beta}}{8\sqrt{2}} - \frac{\sqrt{u^2\beta}}{\sqrt{2}(2+\sqrt{1+4(1-f+2u)})}\right]$$
(25)  
$$P_9 = \frac{1}{\sqrt{-f\beta}} e^{\frac{1}{2}\left(-\frac{f}{8}+\frac{u}{2}\right)\beta - \frac{1}{8}\sqrt{-f\beta}\sqrt{u^2\beta}}\sqrt{2\pi} \left(-\frac{\beta}{16} + \frac{\beta\sqrt{u^2\beta}}{16\sqrt{-f\beta}}\right)$$
(26)

$$P_{10} = (1 + e^{\frac{1}{4}\sqrt{-f\beta}\sqrt{u^2\beta}}(-1 + \text{Erf}[\frac{\frac{(2+\sqrt{1+4(1-f+2u)})^2\sqrt{-f\beta}}{8\sqrt{2}} + \frac{\sqrt{u^2\beta}}{\sqrt{2}}}{2+\sqrt{1+4(1-f+2u)}}]) + \text{Erf}[\frac{(2+\sqrt{1+4(1-f+2u)})\sqrt{-f\beta}}{8\sqrt{2}} - \frac{\sqrt{u^2\beta}}{\sqrt{2}(2+\sqrt{1+4(1-f+2u)})}])$$
(27)

$$P_{11} = \left(\sqrt{2\pi}\beta\left(1 + e^{\frac{1}{4}\sqrt{-f\beta}\sqrt{u^{2}\beta}}\left(-1 + \operatorname{Erf}\left[\frac{\frac{(2+\sqrt{1+4}(1-f+2u))^{2}\sqrt{-f\beta}}{\frac{8\sqrt{2}}{2+\sqrt{1+4}(1-f+2u)}}\right]\right) + \operatorname{Erf}\left[\frac{(2+\sqrt{1+4}(1-f+2u))\sqrt{-f\beta}}{\frac{8\sqrt{2}}{2+\sqrt{1+4}(1-f+2u)}}\right]\right) + \left(28\right)$$



Figure 3: The plot of Magnetisation Under Varying Dimensionless Potential Parameters

Figure 3 presents the variation of magnetisation  $MM(V_o, V_1, \beta)$  at finite temperatures. In Plot (a), magnetisation decreases as the dimensionless potential parameter  $V_o$  increases, particularly with rising values of the temperature-related parameter  $\beta$ , moving from 0.01 to 0.04. This indicates that higher thermal effects suppress magnetisation with increasing  $V_o$ . Plot (b) reveals a consistent trend where magnetisation increases as both the temperature parameter  $\beta$ \beta $\beta$  and the potential parameter  $V_1$  increase.

In Plot (c), magnetisation plotted against  $V_1$  for different values of  $V_0$  shows a positive correlation, with magnetisation increasing as  $V_1$  rises. Conversely, plot (d), which also plots magnetisation against  $V_1$  but under

variations in  $\beta$ , indicates that magnetisation decreases at higher values of  $V_1$ , suggesting a dampening effect from increased thermal activity.

Plot (e) demonstrates that a simultaneous increase in both the temperature parameter  $\beta$  and  $V_1$  leads to a steady rise in magnetisation. Similarly, plot (f) shows that when magnetisation is plotted against  $\beta$ , there is a noticeable increase in magnetisation with rising values of both  $\beta$  and  $V_0$ , highlighting the combined effect of thermal and potential parameters.

#### Magnetic Susceptibility can be express as follows

$$X_m = \frac{\partial M}{\partial B} \tag{29}$$



Figure 4: The plot of Magnetic Susceptibility Under Varying Dimensionless Potential Parameters

Figure 4 illustrates the behaviour of magnetic susceptibility across plots (a) to (f). In Plot (a), a decline in magnetic susceptibility is observed as both the temperature parameter and the dimensionless potential parameter increase. In contrast, plot (b) demonstrates that magnetic susceptibility rises with an increase in both the temperature parameter and the potential parameter  $V_1$ .

Plot (c) shows a growth in magnetic susceptibility when it is plotted against  $V_1$ , while varying the dimensionless potential parameter  $V_0$ , indicating a positive relationship. However, in Plot (d), when both  $V_1$  and  $V_0$  reach higher values, a decrease in susceptibility is observed, suggesting a saturation or suppressive effect at elevated parameter levels.

Finally, Plots (e) and (f) display the variation of magnetic susceptibility with respect to the temperature parameter, under different values of  $V_1$  and  $V_0$ . These plots reveal that increasing the potential parameters leads to a corresponding increase in computed magnetic susceptibility.

The molecular vibrational internal energy is express as  $U = -\frac{\partial}{\partial t} (LogZ)$ 

$$U = -\frac{\partial}{\partial \beta} (LogZ)$$
(30)  
$$U = -(\text{Log}[\beta] / \text{Log}[\frac{1}{\sqrt{-\frac{af\beta\hbar^2}{b\mu}}} (-1 + \text{Erf}[\frac{\frac{P_{12} + \sqrt{\frac{f}{4b\delta^2} + \frac{af}{4b\delta^2} + \frac{u}{\delta}}^2 \delta^2 h^2}{1 + n + \sqrt{1 + 4} (-\frac{f}{2b\delta^2} - \frac{af}{2b\delta^2} + \frac{2u}{\delta} + \lambda}]) + P_{13}$$
(30)

$$P_{11} = e^{-\frac{1}{8}\sqrt{-\frac{af\beta\hbar^{2}}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^{2}} + \frac{af}{4b\delta^{2}} + \frac{u}{\delta})^{2}\delta^{2}\hbar^{2}}{\mu} + \frac{1}{2}\beta(\frac{(-\frac{f}{4b\delta^{2}} + \frac{af}{4b\delta^{2}} + \frac{u}{\delta})\delta^{2}\hbar^{2}}{2\mu} - \frac{af\delta^{2}\hbar^{4}}{8b\mu^{2}})\sqrt{2\pi}(1 + e^{\frac{1}{4}}\sqrt{-\frac{af\beta\hbar^{2}}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^{2}} + \frac{af}{4b\delta^{2}} + \frac{u}{\delta})^{2}\delta^{2}\hbar^{2}}{\mu}}$$
(31)

$$P_{12} = \frac{(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2} - \frac{af}{2b\delta^2} + \frac{2u}{\delta} + \lambda)})^2 \sqrt{-\frac{af\beta\hbar^2}{b\mu}}}{8\sqrt{2}}$$
(32)

$$P_{13} = \operatorname{Erf}\left[\frac{(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)})\sqrt{-\frac{af\beta\hbar^2}{b\mu}}}{8\sqrt{2}} - \frac{\sqrt{\frac{\beta(-\frac{f}{4b\delta^2}+\frac{af}{4b\delta^2}+\frac{u}{\delta})^2\delta^2\hbar^2}{\mu}}}{\sqrt{2}(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)})}\right]$$
(33)

3.4 The molecular vibrational free energy can be express  

$$F = \frac{1}{\beta} (LogZ) = P_{14} + P_{15}$$
(34)

$$P_{14} = \frac{1}{\beta} \log[\frac{1}{\sqrt{-\frac{af\beta\hbar^2}{b\mu}}} e^{-\frac{1}{8}\sqrt{-\frac{af\beta\hbar^2}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^2} + \frac{af}{4b\delta^2} + \frac{u}{b})^2 \delta^2 \hbar^2}{\mu}} + \frac{1}{2}\beta(\frac{(-\frac{f}{4b\delta^2} + \frac{af}{4b\delta^2} + \frac{u}{b})\delta^2 \hbar^2}{2\mu} - \frac{af\delta^2 \hbar^4}{8b\mu^2})\sqrt{2\pi}$$
(35)

$$P_{15} = (1 + e^{\frac{1}{4}\sqrt{-\frac{af\beta\hbar^2}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^2} + \frac{af}{ab\delta^2} + \frac{u}{\delta})^2\delta^2\hbar^2}{\mu}}}(-1 + \operatorname{Erf}\left[\frac{\frac{(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2} - \frac{af}{2b\delta^2} + \frac{2u}{\delta} + \lambda)})^2\sqrt{-\frac{af\beta\hbar^2}{b\mu}}}{\sqrt{2}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^2} + \frac{af}{ab\delta^2} + \frac{u}{\delta})^2\delta^2\hbar^2}{\sqrt{2}}}}{1+n+\sqrt{1+4(-\frac{f}{2b\delta^2} - \frac{af}{2b\delta^2} + \frac{2u}{\delta} + \lambda)}}}\right])$$
(36)





Figure 5: The Plot of Vibrational Internal Energy Under Varying Dimensionless Potential Parameters

Figure 5 illustrates the trends in molecular vibrational internal energy across several plots. In plot (a), it's observed that as both the dimensional potential parameter  $V_1$  and the temperature parameter  $\beta$  increase, the molecular vibrational internal energy tends to decrease. Plot (b) shows a different behaviour: an increase in both parameters leads to a rise in molecular vibrational internal energy. Plot (c) demonstrates that

the internal energy grows as both  $V_1$  and  $V_0$  increase. Conversely, plot (d) reveals that a rise in  $V_1$  and  $V_0$ leads to a decrease in molecular vibrational internal energy. Finally, plots (e) and (f) depict how variations in the temperature parameter affect the vibrational internal energy: specifically, increasing the dimensional potential parameter results in a decrease in molecular vibrational internal energy.





Figure 6: The plot of Vibrational Free Energy Under Varying Dimensionless Potential Parameters

Figure 6 presents the molecular vibrational free energy across different conditions. In plot (a), both the temperature parameter and the dimensionless potential parameter increase, causing the molecular vibrational free energy to rise. However, plot (b) reveals that an increase in these same parameters leads to a reduction in the molecular vibrational free energy. Plot (c) illustrates that as the dimensionless potential parameter  $V_1$  grows, the molecular vibrational free energy decreases. In

contrast, plot (d) shows that higher values of  $V_0$  cause the molecular vibrational free energy to increase. Likewise, plots (e) and (f) demonstrate that when the temperature parameter is plotted against the molecular vibrational free energy, simultaneous increases in  $V_1$ ,  $V_0$ and the temperature parameter  $\beta$  result in a decline in the molecular vibrational free energy.

$$The molecular vibrational entropy can be express as S = k_{\beta} \frac{\partial}{\partial \beta} (F)$$

$$(37)$$

$$P_{16} = \frac{(P_{17}P_{18}P_{19}(-\frac{af(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)})^2h^2}{16\sqrt{2}b\mu} - \frac{(-\frac{f}{4b\delta^2}+\frac{af}{4b\delta^2}+\frac{u}{\delta})^2\delta^2h^2}{2\sqrt{2}\mu} (\sqrt{n}(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)})}) + P_{20}P_{21}+P_{22}+P_{23}+P_{24}}{P_{26}} + P_{27} (38)$$

$$P_{17} = e^{-\frac{1}{8}\sqrt{-\frac{af\beta\hbar^2}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^2} + \frac{af}{4b\delta^2} + \frac{u}{\delta})^2 \delta^2 \hbar^2}{\mu} + \frac{1}{2}\beta(\frac{(-\frac{f}{4b\delta^2} + \frac{af}{4b\delta^2} + \frac{u}{\delta})\delta^2 \hbar^2}{2\mu} - \frac{af\delta^2 \hbar^4}{8b\mu^2})\sqrt{2\pi}}$$
(39)

$$P_{18} = \left(e^{\frac{1}{8}\sqrt{-\frac{af\beta\hbar^2}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^2} + \frac{af}{4b\delta^2} + \frac{u}{\delta})^2\delta^2\hbar^2}{\mu} - \frac{1}{2}\beta(\frac{(-\frac{f}{4b\delta^2} + \frac{af}{4b\delta^2} + \frac{u}{\delta})\delta^2\hbar^2}{2\mu} - \frac{af\delta^2\hbar^4}{8b\mu^2})\sqrt{-\frac{af\beta\hbar^2}{b\mu}}(\frac{1}{\sqrt{-\frac{af\beta\hbar^2}{b\mu}}})$$
(40)

$$P_{19} = (2e^{-\frac{1}{4}\sqrt{-\frac{aff h^2}{b\mu}}\sqrt{\frac{\beta(-\frac{1}{bh^2}+\frac{aff h^2}{bh^2}+\frac{b}{bh^2}\sqrt{\frac{\beta(-\frac{1}{bh^2}+\frac{aff h^2}{bh^2}+\frac{b}{bh^2}+\frac{b}{bh^2}}{\mu}}}{(41)}} \frac{\beta(-\frac{1}{bh^2}+\frac{aff h^2}{bh^2}+\frac{b}{bh^2}+\frac{b}{bh^2}+\frac{b}{bh^2}}{\mu}} (41)$$

$$P_{20} = \frac{1}{\sqrt{\pi}}2e^{-\frac{(1+n+\sqrt{1+4(-\frac{1}{2bh^2}-\frac{aff h^2}{bh^2}+\frac{b}{b$$

$$\sqrt{-b\mu} = 16\mu \sqrt{\frac{\mu^2 - 4b\delta^2 + 4b\delta^2 + \delta^2}{\mu}}$$

$$\tag{47}$$





Figure 7: The Plot of Vibrational Entropy Under Varying Dimensional Potential Parameters

Figure 7 presents the molecular vibrational entropy across various conditions. In plot (a), the entropy exhibits a linear relationship as the dimensionless potential parameter and temperature parameter increase within the range of  $\beta$ =0.01–0.02, but shows a slight decline when  $\beta$ =0.022. Similarly, plot (b) demonstrates that as the temperature parameter  $\beta$  and dimensionless potential parameter  $V_0$  increase together, the molecular vibrational entropy also rises. In plot (c), an increase in both dimensionless potential parameters  $V_0$  and  $V_1$  leads to higher molecular vibrational entropy. In contrast, plot (d) reveals that decreasing  $V_0$  and  $V_1$  results in a reduction in the molecular vibrational entropy. Lastly, plots (e) and (f) show the molecular vibrational entropy as a function of the temperature parameter, highlighting those simultaneous increases in  $V_0$ ,  $V_1$ , and  $\beta$  contribute to an overall rise in the molecular vibrational entropy.

$$\begin{aligned} \text{The molecular vibrational specific heat capacity can express} \\ C_{V} &= k_{\beta} \frac{\partial}{\partial \beta} (U) \end{aligned} \tag{50} \\ P_{28} &= \frac{P_{29}(P_{32}\sqrt{2\pi}P_{30})}{(\sqrt{\pi}(1+n+\sqrt{1+4}(-\frac{f}{2b\delta^{2}}-\frac{af}{2b\delta^{2}}+\frac{2u}{\delta}+\lambda)))} + P_{33} + P_{34} + P_{35}P_{36} + P_{37} + P_{38} + P_{39}P_{40}P_{41} + P_{42} \tag{51} \\ P_{29} &= e^{\frac{1}{8}\sqrt{-\frac{af}{b\mu^{2}}}\frac{A}{b\mu}} \sqrt{\frac{\beta(-\frac{f}{4b\delta^{2}}+\frac{af}{4b\delta^{2}}+\frac{u}{\delta})^{2}\delta^{2}h^{2}}{\mu}} \frac{1}{2}\beta(\frac{-\frac{f}{4b\delta^{2}}+\frac{af}{4b\delta^{2}}+\frac{u}{\delta})^{2}h^{2}}{2\mu} - \frac{af\delta^{2}h^{4}}{8b\mu^{2}}} (52) \\ P_{30} &= ((P_{31} - \frac{af(1+n+\sqrt{1+4}(-\frac{f}{2b\delta^{2}}-\frac{af}{2b\delta^{2}}+\frac{2u}{\delta}+\lambda))^{2}h^{2}}{16\sqrt{2}b\mu\sqrt{-\frac{af\betah^{2}}{b\mu}}}} + \frac{(-\frac{f}{4b\delta^{2}}+\frac{af}{4b\delta^{2}}+\frac{u}{\delta})^{2}\delta^{2}h^{2}}{\mu} )) \\ P_{31} &= 2e^{\frac{1}{4}\sqrt{-\frac{af\betah^{2}}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^{2}}+\frac{af}{4b\delta^{2}}+\frac{u}{\delta})^{2}\delta^{2}h^{2}}{(1+n+\sqrt{1+4}(-\frac{f}{2b\delta^{2}}-\frac{af\delta^{2}}{2b\delta^{2}}+\frac{2u}{\delta}+\lambda))^{2}}\sqrt{\frac{af\betah^{2}}{\mu}}} (\frac{1}{\sqrt{-\frac{af\betah^{2}}{b\mu}}} e^{-\frac{1}{8}\sqrt{-\frac{af\betah^{2}}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^{2}}+\frac{af}{4b\delta^{2}}+\frac{u}{\delta})^{2}\delta^{2}h^{2}}{\mu}} + \frac{1}{2}\beta(\frac{-\frac{f}{4b\delta^{2}}+\frac{af}{4b\delta^{2}}+\frac{u}{\delta})^{2}\delta^{2}h^{2}}{2\mu}} (54) \\ P_{32} &= \sqrt{-\frac{af\betah^{2}}{b\mu}}} \left(\frac{1}{\sqrt{-\frac{af\betah^{2}}{b\mu}}}} e^{-\frac{1}{8}\sqrt{-\frac{af\betah^{2}}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^{2}}+\frac{af}{4b\delta^{2}}+\frac{u}{\delta})^{2}\delta^{2}h^{2}}{\mu}}} \frac{\beta(-\frac{f}{4b\delta^{2}}+\frac{af}{4b\delta^{2}}+\frac{u}{\delta})^{2}\delta^{2}h^{2}}{\mu}} (54) \\ \end{array}$$

$$P_{33} = \frac{1}{\sqrt{\pi}} 2e^{-\left(\frac{(1+n+\sqrt{1+4}-2b\delta^2-2b\delta^2+\delta^{-}(n))}{8\sqrt{2}} - \frac{\sqrt{-4b\delta^2-4b\delta^2}}{\sqrt{2}(1+n+\sqrt{1+4}(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda))}\right)^2}$$
(55)

$$P_{34} = \left(-\frac{af(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)})\hbar^2}{16\sqrt{2}b\mu\sqrt{-\frac{af\beta\hbar^2}{b\mu}}} - \frac{(-\frac{f}{4b\delta^2}+\frac{af}{4b\delta^2}+\frac{u}{\delta})^2\delta^2\hbar^2}{2\sqrt{2}(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)})\mu\sqrt{\frac{\beta(-\frac{f}{4b\delta^2}+\frac{af}{4b\delta^2}+\frac{u}{\delta})^2\delta^2\hbar^2}{\mu}}\right) (56)$$

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$$P_{35} = e^{\frac{1}{4}\sqrt{\frac{affh^2}{b\mu}}\sqrt{\frac{\mu(-\frac{1}{abg^2}+\frac{af}{bhg^2}\sqrt{\mu})^2\beta^2h^2}{g}}} \frac{e^{(\frac{1}{abg^2}+\frac{af}{abg^2}+\frac{af}{b})^2\beta^2h^2\sqrt{\frac{affh^2}{b\mu}}}{g}}} \frac{e^{h^2}\sqrt{\frac{\mu(-\frac{1}{abg^2}+\frac{af}{abg^2}+\frac{af}{b})^2\beta^2h^2}{g}}} \frac{e^{-\frac{1}{4}h^2}\sqrt{\frac{\mu(-\frac{1}{abg^2}+\frac{af}{abg^2}+\frac{af}{b})^2\beta^2h^2}{g}}} \frac{e^{-\frac{1}{4}h^2}\sqrt{\frac{affh^2}{b\mu}}}{g}} \frac{e^{-\frac{1}{4}\sqrt{\frac{affh^2}{abg^2}+\frac{af}{abg^2}+\frac{af}{abg^2}+\frac{af}{b}}} \frac{e^{-\frac{1}{4}\sqrt{\frac{affh^2}{abg^2}+\frac{af}{abg^2}+\frac{af}{abg^2}+\frac{af}{b}}}}{g}} \frac{e^{-\frac{1}{4}\sqrt{\frac{affh^2}{abg^2}+\frac{af}{abg^2}+\frac$$

$$P_{44} = \operatorname{Erf}\left[\frac{(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2} - \frac{af}{2b\delta^2} + \frac{2u}{\delta} + \lambda)})\sqrt{-\frac{af\beta\hbar^2}{b\mu}}}{8\sqrt{2}} - \frac{\sqrt{\frac{f(-\frac{f}{4b\delta^2} + \frac{af}{4b\delta^2} + \frac{u}{\delta})^2\delta^2\hbar^2}{\mu}}}{\sqrt{2}(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2} - \frac{af}{2b\delta^2} + \frac{2u}{\delta} + \lambda)})}\right]$$
(65)

55



$$P_{46} = \frac{\frac{1}{\sqrt{2\pi}P_{47} - \frac{1}{\frac{\sqrt{2(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2} + \frac{af}{2b\delta^2} + \frac{u}{\delta})^2 \delta^2 \hbar^2}}{\mu}}}{\sqrt{2}(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2} - \frac{af}{2b\delta^2} + \frac{2u}{\delta} + \lambda})})k_{\beta}}$$
(67)

$$P_{47} = \left(-1 + \operatorname{Erf}\left[\frac{\frac{(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)})^2}{8\sqrt{2}}\sqrt{-\frac{af\beta\hbar^2}{b\mu}}}{1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)}}\right] \left(1 + e^{\frac{1}{4}\sqrt{-\frac{af\beta\hbar^2}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^2}+\frac{af}{4b\delta^2}+\frac{u}{\delta})^2\delta^2\hbar^2}{\mu}}}{1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)}}}\right) + \operatorname{Erf}\left[\frac{(1+n+\sqrt{1+4(-\frac{f}{2b\delta^2}-\frac{af}{2b\delta^2}+\frac{2u}{\delta}+\lambda)})}{8\sqrt{2}}\right] \left(1 + e^{\frac{1}{4}\sqrt{-\frac{af\beta\hbar^2}{b\mu}}\sqrt{\frac{\beta(-\frac{f}{4b\delta^2}+\frac{af}{4b\delta^2}+\frac{u}{\delta})^2\delta^2\hbar^2}{\mu}}}{(68)}\right)$$



56



Figure 8: The plot of vibrational entropy under varying dimensionless potential parameters.

Figure 8 depicts the molecular vibrational specific heat capacity as a function of various parameters. In plot (a), increasing the temperature parameter in the range of  $\beta = 0.01$  to 0.02 causes an increase in the specific heat capacity, while at  $\beta = 0.03$ , the capacity decreases. Plot (b) demonstrates that a simultaneous increase in the temperature parameter and the dimensionless potential parameter  $V_0$  leads to a higher specific heat capacity. Plot (c) reveals that an increase in both dimensionless potential parameters,  $V_0$  and  $V_1$ , results in a reduction of the specific heat capacity. In contrast, plot (d) shows that an increase in the dimensionless potential parameter alone leads to an increase in the specific heat capacity.

Plots (e) and (d) also plot the molecular vibrational specific heat capacity against the temperature parameter, further confirming that an increase in the dimensionless potential parameter results in a higher specific heat capacity.

Magnetisation and magnetic susceptibility at Zero Temperature can be writen as

$$M_{nm} = -\frac{\partial(E_{nm})}{\partial V_0} \tag{69}$$

$$X_M = \frac{\partial M_{\rm nm}}{\partial V_0} \tag{70}$$







Figure 9: The Plot Magnetisation at Zero Under Varying Dimensionless Potential Parameters

Figure 9 displays the plot of magnetization at zero temperature for varying dimensionless potential parameters. Each subplot illustrates how magnetization and magnetic susceptibility depend on the dimensionless potential parameters  $V_0$  and  $V_1$ . In plot (a), the magnetization  $M_{mm}(V_0, V_1)$  is shown as a function of  $V_0$  for fixed values of  $V_1$ . When  $V_1=0$ , magnetization initially increases with  $V_0$  before dropping sharply at larger  $V_0$ . For higher values of  $V_1$  (e.g., 5 and 10), magnetization increases more slowly and then also decreases at higher  $V_0$ . This indicates that while magnetization is enhanced at small  $V_0$ , it is suppressed at larger  $V_0$ , with the suppression becoming more pronounced as  $V_1$  increases.

In plot (b), magnetization is plotted against  $V_1$  for various fixed  $V_0$  values. The magnetization decreases rapidly with increasing  $V_1$ , and the curves nearly coincide, suggesting minimal dependence on  $V_0$ . Plot (c) presents magnetic susceptibility  $\chi_m(V_0, V_1)$  versus  $V_1$ for different  $V_0$  values. All curves display a pronounced dip at low  $V_1$ , followed by a gradual recovery as  $V_1$ grows. Overall, the magnitude of susceptibility remains negative, indicating a diamagnetic response. In plot (d), the susceptibility initially rises with increasing  $V_1$ , peaks, and then decreases at higher  $V_1$ . Notably, the peak location shifts to lower  $V_1$  values as  $V_0$  increases.

## CONCLUSION

This study employs partition function methods derived from the non-relativistic Schrödinger equation to determine thermodynamic properties, emphasizing how dimensionless potential parameters influence the quantum energy spectrum. It finds that energy eigenvalues are highly sensitive to these potential parameters, with the dimensionless parameter  $\lambda$  playing a crucial role in defining the quantum states. Using the calculated energy spectrum, thermodynamic quantities like magnetization and magnetic susceptibility are derived, revealing that magnetization generally decreases with increasing potential parameter  $V_0$  and temperature, but rises with  $V_1$  and higher temperatures. Magnetic susceptibility shows varying trendsdiminishing with higher temperature and  $V_0$ , increasing with  $V_1$ , and saturating at high parameter ranges. Molecular vibrational energy and free energy demonstrate complex dependencies on temperature and potential parameters, with vibrational energy either decreasing or increasing depending on parameter ranges, and vibrational free energy consistently declining when both parameters increase. Entropy grows with higher potential and temperature parameters, reflecting an increase in system disorder, while specific heat capacity varies depending on the conditions, highlighting quantum transitions. At zero temperature, magnetization and magnetic susceptibility exhibit a delicate interplay between  $V_0$  and  $V_1$ , with magnetization showing initial growth and subsequent suppression, and susceptibility displaying characteristic diamagnetic features. The findings underscore the efficacy of using partition function approaches to connect quantum mechanical details to macroscopic thermodynamic behaviour, offering valuable insights for condensed matter physics and materials research.

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